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(54) ARTICLES HAVING IMPROVED SLIP COATINGS

GEGENSTAND MIT GLEITBESCHICHTUNG ARTICLES A REVETEMENTS AUTOGLISSANTS AMELIORES

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Description

This invention relates to rubber and vinyl articles having improved slip coatings and to compositions and methods of providing same. More particularly, the invention relates to rubber and vinyl articles, such as sterile surgeons' gloves, which are provided with an improved slip coating.

It has previously been proposed to provide slip finishes on rubber and vinyl articles such as rubber gloves or girdles by various methods. For example, the surface of a rubber glove can be halogenated with bromine or chlorine to make it slippery. However, this treatment may result in very poor aging properties. Discoloration can begin almost immediately and, within a month, the halogenated surface may become hard and brittle and brown in color. This can be avoided only by taking great care in the halogenation process and even then there is no assurance of obtaining a uniform, sustained, slip film. Waxes and silicone have been used but these provide only a temporary solution as these materials rub off in a very short time. It has further been proposed in U.S. Patent No. 3,286,011 issued March 18, 1964, and U.S. Patent No. 3,411,982 issued November 1968, to provide a slip-finish-comprising a rubber latex and a resin latex. While such coatings reduce the coefficient of friction of the rubber article to a slight extent, it is desirable to further reduce the coefficient of friction. For example, it is desirable to further reduce the coefficient of friction to make it easier to put on and take off a rubber article such as a rubber glove.

In the preparation of rubber articles by dipping into a coagulant solution and then into a rubber latex followed by coagulation of the rubber latex into the desired article, it is usually necessary to first apply to the form a coating of a release agent such as a mold release powder, e.g., talc, diatomaceous earth, etc. or a lubricant type release agent, e.g., glycerine. The reasons for the use of the release agents are (1) to prevent damage to the rubber article when it is stripped from the form and/or (2) to preclude the tendency of the tacky rubber to self adhesion when the article is removed. Use of release agents in the preparation of dipped rubber articles, however, is not without its shortcomings for it naturally leads to adulterated final products which have trapped or otherwise picked up the release agent on the coagulated rubber surface. Nevertheless, the contaminated rubber article is a matter which manufacturers of such articles have had to live with for it has not been possible to obtain these rubber articles in a commercially acceptable form without the use of the release agents.

Another drawback commonly associated with the use of release agents in the manufacture of rubber articles is that the process requires clean up of the residue of the release agent on the mold or form after formation of each and every article, a tedious and time consuming operation.

The same problems necessitating the use of release agents in the formation of rubber articles directly on dip-

ping forms, likewise necessitate the use of release agents in processes such as described in U.S. Patent No. 3,411,981 to Kavalir, et al. wherein the slip-coated rubber articles are prepared by first providing a release composition onto the form before it is dipped into the rubber latex. In addition to the aforementioned stripping and self-adhesion difficulties, there is a tendency for the slip coating to separate from the rubber substrate when release agents are not utilized in such processes.

Furthermore, in the prior art processes for obtaining slip coatings there has been a tendency for the surface of the articles produced to exhibit streaks.

The foregoing problems were overcome or alleviated by use of the two component synthetic polymer systems described in U.S. Patent Nos.4,027,060 and 4,082,862. While rubber articles such as gloves produced using the dual polymer component systems and processes of these patents were excellent, they were not without shortcomings. For one, the entire process takes approximately three hours, a process time considered too long by many. Secondly, the removal of the finished gloves from the forms had to be accomplished by stripping under warm water, an inconvenience that further required yet another final drying step. Thirdly, in the prior art processes, coating of the rubber surface with the polymer systems could not be accomplished when the rubber was cured or partially cured. The coating had to be conducted while the rubber was still hydrated. Lastly, it was difficult, if not impossible, to don the finished glove when the hands are wet or damp.

DE-A-2 230 435 discloses a method for making a slip coating on a rubber surface. An aqueous synthetic latex is used, consisting of water and a mixed polymer of vinyl chloride (or vinylidene chloride) and ethylacrylate.

Thus, it is an object of the invention to provide rubber articles containing slip coatings comprised of non-elastomeric materials which when applied to the rubber substrate take on the elastomeric properties of the substrate.

It is also an object of the invention to provide a process for the production of rubber articles which process does not require the use of release agents as a separate and distinct entity and which nevertheless results in a rubber article having a slip coating which has an appreciably lower coefficient of friction than the rubber surface to which it has been applied.

Yet another objective of the invention is to provide a process for the production of rubber articles substantially reduced in process time where such process time is commensurate with or faster than existing commercial cycles.

Another object of the present invention is to provide a process that enables removal of the finished glove from the forms without having to immerse them in warm water.

Still another object of the invention is to provide a process for the manufacture of rubber articles such as gloves which permits coating over cured, partially cured, or uncured (hydrated) rubber and wherein a second deionized leach step is unnecessary.

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Yet another object of the invention is to provide rubber or vinyl articles such as gloves which can be donned with wet or damp hands.

BRIEF SUMMARY OF THE INVENTION

The foregoing and other objects of the invention will be apparent to those having ordinary skill in the art and are achieved according to the present invention by providing a formed article comprising outer and inner rubber or vinyl resin surfaces, at least one of said outer and inner rubber or vinyl surfaces having a first essentially rubberfree extensible adherent slip coating comprising a filmforming copolymer resin Component A comprising the reaction product of 55 to 80 mole %, preferably 60 to 70 mole % of at least one copolymerizable vinyl halide, 15 to 32 mole %, preferably 20 to 27 mole %, of at least one copolymerizable ethylenically unsaturated hydrocarbon and 3 to 10 mole %, preferably 4 to 8 mole % of at least one copolymerizable monomer containing a carboxyl or amido group or a second essentially rubber-free extensible adherent slip coating comprising a film forming copolymer resin Component B comprising the reaction product of 76 to 94 mole %, preferably 80 to 90 mole % of a copolymerizable vinyl or vinylidene halide, 4 to 12 mole %, preferably 8 to 11 mole % of acrylonitrile and 2 to 12 mole %, preferably 2 to 6 mole % of at least one acrylic monomer, or mixtures of Component A and Component B.

In a preferred embodiment, the invention provides a formed article comprising outer and inner rubber or vinyl resin surfaces, said outer rubber or vinyl surface having a first essentially rubber-free extensible adherent slip coating comprising a film-forming copolymer resin Component A comprising the reaction product of 55 to 80 mole %, preferably 60 to 70 mole % of at least one copolymerizable vinyl halide, 15 to 32 mole %, preferably 20 to 27 mole %, of at least one copolymerizable ethylenically unsaturated hydrocarbon and about 3 to 10 mole %, preferably 4 to 8 mole % of at least one copolymerizable monomer containing a carboxyl or amido group, and said inner rubber or vinyl surface having a second essentially rubber-free extensible adherent slip coating selected from the group consisting of Component A, a film forming copolymer resin Component B comprising the reaction product of 76 to 94 mole %, preferably 80 to 90 mole % of a copolymerizable vinyl or vinylidene halide, 4 to 12 mole %, preferably 8 to 11 mole % of acry-Ionitrile and 2 to 12 mole %, preferably 2 to 6 mole % of at least one acrylic monomer, or mixtures of Component A and Component B.

In another preferred embodiment of the present invention the first slip coating, i.e., the coating on the outer surface, includes a detackifying material. A particularly preferred detackifying material comprises the copolymer resin Component B identified above.

In the preferred embodiment of the present invention, Component B is used on the inner surface; however, with certain rubber substrates, it may be necessary to use either Component A alone or a mixture of Component B and Component A to obtain satisfactory adhesion of the coating to the inner surface of the rubber or vinyl. In such cases a mixture of Component A and Component B is preferred.

In yet another embodiment of the present invention, the second slip coating provided on the inner surface of the article includes a small amount of an alkali metal, alkaline earth metal, ammonium or trialkanolamine salt of a fatty acid of 8 to 24 carbon atoms. It has been found that this material permits coating over a dry or partially cured rubber or fused vinyl during the manufacturing process.

Further, by adjusting the pH of the base resin emulsion system (i.e. either Component A or Component B or mixtures thereof) anionic, cationic or nonionic surfactants can be used to effect coating over cured, partially cured or uncured (hydrated) rubber.

Aside from the salts of fatty acids described above, other anionic surfactants found effective include the sulfates and sulfonates of oils and fatty acids - for example, sodium dodecylsulfate, sulfated glycerol trioleate, sodium alkylaryl sulfonate, dodecylbenzene sulfonic acid. The esters of sodium sulfosuccinic acid are also useful.

In the nonionic class, octylphenoxypolyethoxyethanol is found to be most effective. Cationics such as tallow triammonium chloride and dicocodimethylammonium chloride have been utilized.

Another preferred embodiment of the invention includes in combination with copolymer resin Component B, a small amount of a poly(alkylene oxide) glycol. It has been found that the presence of this material in combination with the second copolymer resin component provides a dry lubricating coating that offers better wet-hand donnability than when copolymer resin Component B alone is employed.

The use of a polyglycol and a fatty acid salt produces a gel which functions to lubricate the coating and impart a velvety feel to the skin. Gels of varying consistency are produced by the other surfactants depending on functionality.

In another aspect of the invention, there is provided an emulsion of copolymer B containing small amounts of a poly(alkylene oxide) glycol and optionally containing a small amount of a silicone fluid. The silicone can be polydimethylsiloxane or an aminopolysiloxane. Adding silicone in an amount of 2 to 15% by weight of total solids further improves damp hand donnability while maintaining gel properties. The result is an aesthetically pleasing dry lubricating coating with damp hand donnability in contrast to fluid coatings which produce greasy or oily films on the inside of the glove.

The novel process of the invention comprises coating a form with an aqueous dispersion of copolymer resin Component A or Component B as defined above, or mixtures thereof, preferably a coating comprising Component A to form a first, rubber-free extensible adherent slip coating, forming over said slip coating a rubber or vinyl

substrate, at least partially curing the rubber or fusing the vinyl. In a preferred embodiment, the thus formed substrate is then coated with an aqueous dispersion of the copolymer resin Component A or Component B defined above, or a combination of Component A and Component B, and the composite thus formed is dried and stripped off of the form to provide rubber or vinyl articles with improved slip coatings on both the outer and inner surfaces of the article. In the case of surgeons' gloves, when the outer surface is coated with a coating comprising Component A there is thereby produced a glove which can be easily stripped off of the form without submersion in warm water, which is free of any release agent contamination, which can be slipped on and off even when the hands are wet or damp.

In another aspect of the invention there is provided an aqueous dispersion of copolymer resin Component A defined above and detackifying amounts of copolymer resin Component B, optionally containing small amounts of a silicone.

In yet another aspect of the invention there is provided an aqueous dispersion of copolymer resin Component B containing small amounts of an alkali metal, alkaline earth metal, ammonium or trialkanolamine salt of a fatty acid of 8 to 24 carbon atoms, optionally-con--25 taining a small amount of poly(alkylene oxide) glycol, and a small amount of polydimethylsiloxane or an aminopolysiloxane.

DETAILED DESCRIPTION OF THE INVENTION

Rubber or Vinyl Surface

The rubber surface on which the slip coating is provided may be fabricated from any suitable conventional latex dipping compound such as those disclosed in U.S. Patent No. 3,411,982. The latex may be pre-cured or non pre-cured latex in which case the rubber article is cured after forming. The latex dip may contain conventional compounding ingredients commonly utilized. Specific examples are given in U.S. Patent No. 3,411,982. The rubber may be natural rubber or any conventional latex suitable for dipping operations. Of the various natural and synthetic latices, natural rubber, polychloroprene rubber, synthetic polyisoprene, SBR, and mixtures thereof are preferred.

Vinyl surfaces may be prepared from vinyl resins such as polyvinylchloride resin using conventional processes well known in the art.

Conventional formulations for these rubber and vinyl resins are well known in the art and those skilled in the art are readily able to vary the formulations and conditions of curing or fusing and the like to suit the particular latex being used as well as the particular final article desired. Similarly, the articles can vary widely and include gloves, particularly surgeons' gloves, girdles, and the like.

First or Outer Slip Coating

Copolymer resin Component A preferably employed in providing the first slip coating on the outer surface is comprised of the reaction product of at least one copolymerizable vinyl halide, at least one copolymerizable ethylenically unsaturated hydrocarbon and at least one copolymerizable monomer containing a carboxyl or an amido group, in the proportions defined above.

Illustrative of vinyl halides that may be employed are vinyl chloride, vinyl fluoride and vinyl bromide. The preferred vinyl halide is vinyl chloride.

The ethylenically unsaturated hydrocarbon monomers of the copolymer resin Component A are preferably straight and branch-chained alpha-olefins of 2 to 5 carbon atoms such as ethylene, propylene, 1-butene, isobutene, 1-pentene and the like.

The copolymerizable ethylenically unsaturated monomers containing at least one carboxyl or amido group of the copolymer resin Component A are preferably acrylic monomers having the structure:

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X is -OH, NH₂,

OCH₃, OC₂H₅ or OCH₂CH₂CH₂CH₃. R is an H or methyl.

Examples of suitable acrylic monomers are acrylic acid, methacrylic acid, and esters of acrylic acid and methacrylic acid preferably up to 12 carbon atoms such as methyl acrylate, ethyl acrylate, n-butyl acrylate, isobutyl acrylate, 2-ethylhexyl acrylate, lauryl acrylate, and similarly methyl methacrylate, ethyl methacrylate, n-butyl methacrylate, isobutyl methacrylate, 2-ethylhexyl methacrylate, lauryl methacrylate, acrylamide, methacrylate, lauryl methacrylate, acrylamide, methacrylamide or N-methylolacrylamide; and mixtures thereof.

The preferred ethylenically unsaturated hydrocarbon monomer is a mixture of acrylamide and acrylic acid in a ratio, preferably between 1:1 and 3:1.

It has been found that copolymer resin Component A is essential for proper release of the substrate from the form and permits release from a bisque form without the necessity of carrying out the removal while submerged under warm water. Until copolymer resin Component A was used in the outer slip coating it was impossible to strip from a bisque form. Heretofore, prior art systems required glazed porcelain forms to produce a powderless

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glove with the very smooth surface of the glazed porcelain form facilitating removal from the form. However, even with the glazed forms, removal had to be effected under water. The ease of release from bisque forms is surprising since the polar groups present in copolymer resin Component A are expected to interfere with good release.

Thus, where Component B is used alone as the outer slip coating it will be necessary to effect removal from porcelain forms while submerged under water pursuant to prior art techniques as described, for example, in U.S. Patent Nos. 4,027,060 and 4,082,862.

When employed the detackifier component of the first slip coating is present in detackifying amounts generally falling in the range of 10 to 25% by weight of the total solids.

The detackifier component preferably used in combination with the first resin component can be any of the detackifier resins conventionally used for this purpose such as waxy materials, for example, cetyl alcohol, stearic acid esters and polyethylene emulsions. Preferred, however, is a copolymer resin Component B comprising the reaction product of vinylidene or vinyl halide, acrylonitrile and an acrylic monomer. By "acrylic monomer" as used in this specification and the appended claims, is meant a copolymerizable acrylic acid, lower alkyl acrylic acid and the esters of acrylic acid and lower alkyl acrylic acid.

The silicone component, optionally included in the first or outside slip coating, can be any of the well known polysiloxanes commonly referred to as silicones and preferably comprised of organopolysiloxane, a dialkylsiloxane polymer, or an aminopolysiloxane. Especially preferred is dimethylsiloxane polymer. It has been found that the presence of small amounts of the polysiloxane induces suppleness to the finished rubber article. When used, therefore, it is added in an amount that increases the suppleness of the outer slip coating. In general, these amounts fall in the range of 0.001 to 0.5% by weight of the total solids. If preferred, this silicone component can be formulated into the rubber latex also to further add suppleness to the end product.

Second or Vinyl Inner Slip Coating

If a second or inner slip coating is applied it can be either Component A described above but is preferably Component B. The copolymer resin Component B of the second slip coating comprises the reaction product of a copolymerizable vinyl or vinylidene halide, acrylonitrile and acrylic monomer in the portions defined above. As aforementioned, copolymer resin Component B of the second slip coating has been found to provide an inner coating on rubber or vinyl gloves that facilitates the donning of the gloves when the hands are wet or damp. Preferred amongst this class of copolymer resins is the copolymer vinylidene halide, acrylonitrile and methylmethacrylate.

It is preferred to include as part of the inner slip coating an alkali metal, alkaline earth metal, ammonium or trialkanolamine salt of a fatty acid having 8 to 24 carbon atoms. The preferred fatty acid salts are the alkali metal or triethanolamine salts of fatty acids of 12 to 20 carbon atoms. Especially preferred is cesium stearate, cesium myristate, cesium palmitate and triethanolamine stearate, myristate or palmitate. The presence in the slip coating of preferably 5 to 15% by weight (solids basis) of these fatty acid salts has been found to permit coating over dried or partially cured rubber. This eliminates the most critical part of the prior art processes which require thorough washing of the coagulated rubber in deionized water after a standard leach to remove any residual divalent ions such as Ca++ and then polymer coating the rubber surface while the rubber was relatively dry but still hydrated. In the prior art processes if the rubber was too wet or too dry (partially-cured) polymer pick-up and therefore coating was impossible. Thus, the fatty acid salt component of the invention permits coating over dry or partially-cured rubber. As mentioned previously, by adjusting the pH of the emulsified resin Component B (or mixtures of resin component A & B) either anionic, cationic or nonionic surfactants can be used to effect coating over cured or uncured rubber.

Even better wet-hand donnability is obtainable if there is included in the second slip coating a polyhydroxyfunctional polymer comprising poly(alkylene oxide) polyol. Not all the alkylene units need be the same. Poly(alkylene oxide) polyols formed by the copolymerization or condensation of mixtures of different cyclic ethers, glycols, or glycols and cyclic ethers can be used; as can poly(alkylene oxide) polyols derived from cyclic ethers such as dioxolane, which affords a polyol having the formula HO(CH2-OCH2CH2O)nH, where n is greater than 1. The alkylene segment can be a straight or a branched chain, as in poly(propylene oxide) polyol. In the case where the alkylene unit is ethylene, it can be advantageous to incorporate the unit into a copolymer, for example, as a copolymer of ethylene oxide and propylene oxide, with up to 80 percent of such copolymer comprising ethylene oxide. Representative poly(alkylene oxide) polyols include poly(ethylene oxide) polyols, poly(propylene oxide) polyols, poly(tetramethylene oxide) polyols, poly(nonamethylene oxide) polyols, poly(oxymethyleneethylene oxide) polyols, poly(ethylene oxide-propylene oxide) polyols, and poly(pentaerythritolethylene oxide) polyols. Thus, the poly(alkylene oxide) polyols will generally have from 2 to 6 hydroxyl groups with such polyols having 2 hydroxyl groups being currently preferred. Preferred poly(alkylene oxide) polyols are poly(ethylene oxide) polyols, poly(propylene oxide) polyols, and poly(ethylene oxide-propylene oxide) polyols.

Also functional in this application are polypropylene glycol esters, polyethyleneglycol esters and polyglycerol esters. Polyethyleneimine resins have also been used successfully.

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When utilized, the poly(alkylene oxide) polyol is employed in amounts ranging from 5 to 50% by weight (solids basis), preferably 5 to 30%. Excessive concentrations of poly(alklene oxide) tend to induce an undesirable oiliness to the finished product.

Thus, while use of the second copolymer resin by itself improves wet-hand donnability, even better results are achieved when poly(alkylene oxide) polyol is employed in combination with the resin. Of interest is the fact that even when liquid poly(alkylene oxide) polyols are used in combination with the second copolymer resin, surprisingly the end product is a dry coating of uniform lubrication even though a film of pure second copolymer resin and the polyglycol used are incompatible. Although the reason for this phenomena is not known for certain, one explanation is that the use of a polyglycol and a fatty acid salt produces a gel which is trapped within the copolymer resin Component B as the film coalesces producing a dry appearance. The polyglycol/fatty acid salt gel is released in minuscule amounts as the glove is stressed, for example, during donning - thereby providing the lubricating function and yet always appearing dry and aesthetically pleasing. Another explanation is an ester interchange reaction with the methylmethacrylate group of the second copolymer-resin which results - 25in a chemical linkage between the resin and the glycol. Adding 2-15% by weight of total solids of silicone further improves damp hand donnability while maintaining gel properties. That gel properties are maintained is postulated on the basis of silicone fluid reaction through the OH function producing pendant silicone groups on the glycol. Further, amino functional silicones can react with the acid or ester component of the second resin system thereby attaching to the base resin by primary bonding forming a branched polymer/polymer alloy in situ.

The above reactions would occur during the precure cycle as the water of dispersion is evaporating and the polymer film is coalescing The result is a cosmetically-appealing, dry lubricating coating.

The improved slip coatings according to the invention are preferably provided on a rubber or vinyl surface by contacting the surface with an aqueous dispersion consisting essentially of water and the copolymer resins. Alternatively, the slip coating can be first provided on a form and the form subsequently dipped into a rubber latex as disclosed in U.S. Patent No. 3,411,982.

The slip coating provided by the present invention has excellent characteristics of slipperiness. A simple way to demonstrate the slipperiness of substrates coated according to the invention is to place two coated surfaces together and rub the surfaces back and forth while grasped between the fingers. Previous slip coatings which have been tested slip poorly when held in this manner. However, the coatings provided by the present invention show a marked slip. Furthermore, the coatings provided by the present invention are tenaciously adhesive and the rubber substrate can be stretched to maximum extent without affecting the slipperiness of and

without causing cracking of the slip coating according to the present invention.

According to a preferred embodiment of the present invention a form for the desired article is preferably heated and provided as by dipping with an aqueous dispersion consisting essentially of water and the copolymer resin Component A. The resin coating is dried on the form and provided with a coagulant such as a solution of calcium nitrate and isopropanol after which it is dipped into a tank containing rubber latex solution. The rubber latex solution generally varies from 25 to 50% solids depending upon the thickness and the viscosity of the desired product. The immersion time into the tank will vary depending on the desired thickness and in general the dip time will extend from about 3 seconds to 1 minute. After the formation of the rubber substrate the coagulant therein is leached, for instance, by dipping the resulting form into a leach tank containing hot water which is usually at a temperature ranging from about 135° to 140° F (57.2 to 60°C)

The article is then partially cured by heating at approximately 200° F (93.2°C) for 15 minutes. In this instance there results a product the outer surface of which contains the first resin slip coating. The second slip coating of the invention is then applied to the partially cured rubber substrate as an aqueous dispersion consisting essentially of water and copolymer resin Component B. The second resin slip coating is then dried and the resin-rubber-resin composite is stripped off the form.

The following examples are included to further illustrate the process of the present invention.

EXAMPLE 1

A bisque glove form is cleaned by pointing it down in a tank of M-Pyrol and soaking for five minutes at room temperature. The form is then later rinsed and dried. The dried form is then dipped into a first polymer coating system of 5% solids and 95% water. Of the 5% solids, 85% is comprised of a) an emulsion at 49% solids of:

Vinylchloride	69%
Ethylene	25%
Acrylamide	4%
Acrylic Acid	2%

15% of b) an aqueous dispersion at 54% solids of:

Vinylidene chloride	85%
Acrylonitrile	10%
Methylmethacrylate	4.5%

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and 0.025% silicone emulsion (a non-ionic emulsion containing 35% dimethylpolysiloxane).

The dip coated form is baked in an oven with the form pointing up at 225° F (107.2°C) for 2 to 3 minutes. The coated glove form is dipped with the form pointing 5 down into a rubber coagulant (calcium nitrate) maintained at 110° F (43.4°C) and then into an aqueous natural rubber latex (35% solids) for 20 to 30 seconds. At the point when the form is completely out of rubber latex, and no further dripping takes place, the form is rotated so that the fingertips are in the up position to distribute the last drop of rubber and subjected to a conventional leaching step at approximately 150° F (65.6°C) for 15 minutes followed by oven curing at approximately 200° F (93.2°C) for 15 minutes. The form containing the first polymer system rubber composite coating is then dipped into a second polymer system composed of 6% solids and 94% water. 5% of the solids is aqueous dispersion (b) to which is added 0.5% by weight cesium stearate and 0.5% by weight polypropylene glycol. The thus coated glove is dried with the fingers up at 205° F (96°C) for 1 minute and then stripped from the mold. Note: the mold is now ready for further glove making without the necessity of cleaning - simply dip into resin Component A and start the cycle all over again.

EXAMPLE 2

Example 1 above is repeated except after the leach the rubber is precured for approximately 5 minutes at 200° F (93.2°C). The form containing the first polymer system rubber composite is now dipped into the second polymer system described above and then cured in an oven for the remaining 10 minutes at 200° F (93.2°C).

EXAMPLE 3

Example 1 above is repeated except along with the cesium stearate and polypropylene glycol gel a silicone is added in an amount of approximately 10% by weight of total solids. The silicone can be polydimethylsiloxane or amino-polysiloxane. This amount of silicone incorporated does not interfere with the lubricating gel produced by the cesium stearate and polyglycol. Further, the amine function of the polysiloxane can react with the acid or ester component of the second resin system thereby attaching itself to the base resin by primary bonding forming a polymer alloy.

EXAMPLE 4

A bisque glove form is cleaned by pointing it down in a tank of M-Pyrol and soaking for five minutes at room temperature. The form is then rinsed and dried. The dried form is then dipped into a first polymer coating system of 6% solids and 94% water. The 6% solids is resin Component A which consists of an emulsion of the following:

Vinylchloride	69%
Ethylene	25%
Acrylamide	4%
Acrylic Acid	2%

Additionally, 0.01 to 0.3% by weight of polypropyene glycol or 0.025% by weight of a silicon emulsion or both can be added as detackifiers. The dip coated form is heated in an oven with the form pointing up at 225°F (107.2°C) for 2 to 3 minutes. The coated glove is dipped with the form pointing down into a rubber coagulant maintained at 110°F (43.4°C) and then into an aqueous natural rubber latex (35% solids) for 20 to 30 seconds. At the point when the form is completely out of rubber latex, and no further dripping takes place, the form is rotated so that the fingertips are in the up position to distribute the last drop of rubber and subjected to a conventional leaching step at approximately 150°F (65.6°C) for 15 minutes followed by oven drying of the surface water (approximately 30 seconds). The form containing the first polymer system rubber composite coating is then dipped into a second tank of Component A resin system at 6% solids and 94% water. In this second dip tank, of the 6% solids, 4.5% is resin Component A described above to which is added 0.75% solids by weight of cesium stearate and 0.75% solids by weight of polypropylene glycol. The thus coated glove is cured with the fingers up at 200°F (93.2°C) for 10-15 minutes, cooled and stripped from the mold. The mold is now ready for further glove making without the necessity of cleaning simply dip into resin Component A and start the cycle all over again.

EXAMPLE 5

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Example 4 is repeated except that along with the cesium stearate and polypropylene glycol gel, a silicone is added in an amount of approximately 10% by weight of total solids. The silicone can be the same as that employed in Example 3 above.

EXAMPLE 6

A bisque glove form is cleaned by pointing it down in a tank of M-Pyrol and soaking for five minutes at room temperature. The form is then rinsed and dried. The dried form is then dipped into polymer coating system B at 6% solids and 94% water. The 6% solids is resin Component B which consists of an aqueous dispersion of the following:

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Vinylidene chloride	85.0%
Acrylonitrile	10.0%
Methylmethacrylate	4.5%

Additionally, 0.01 to 0.3% by weight of polypropylene glycol or 0.025% by weight of silicone emulsion can be added.

The dip_coated form is heated in an oven with the form pointing up at 225°F (107.2°C) for 2 to 3 minutes. The coated glove is dipped with the form pointing down into a rubber coagulant maintained at 110°F (43.4°C) and then into an aqueous natural rubber latex (35% solids) for 20 to 30 seconds. At the point when the form is completely out of the rubber latex, and no further dripping takes place, the form is rotated so that the fingertips are in the up position to distribute the last drop of rubber and subjected to a conventional leaching step at approximately 150°F (65.5°C) for 15 minutes followed by oven drying of the surface water (approximately 30 seconds). The form containing polymer coating system B rubber composite is then dipped into a second tank of polymer 25 coating system B at 6% solids and 94% water. In this second dip tank, of the 6% solids, 4.5% is resin Component B described above to which is added 0.75% solids by weight of cesium stearate and 0.75% by weight of polypropylene glycol. The then coated glove is cured with the fingers up at 200°F (93.2°C) for 10-15 minutes. Note: in this case with resin component system B on both sides, the glove will have to be stripped hydrated, e.g., submerged in a warm water bath for 10 minutes, removed from the bath and subsequently stripped and dried. The mold, however, is ready for further glove making without the necessity of cleaning - simply dip into resin Component system B and start all over again.

EXAMPLE 7

Example 6 is repeated except that along with the cesium stearate and polypropylene glycol gel a silicone is added in an amount of approximately 10% by weight of total solids. The silicone can be the same as employed in Example 3 above.

Claims

1. A formed article comprised of outer and inner rubber 50 or vinyl resin surfaces, at least one of said outer and inner surfaces having an essentially rubber-free extensible adherent slip coating selected from the group consisting of (i) a first film-forming copolymer resin Component A comprising the reaction product 55 of 55 to 80 moles % of at least one copolymerizable vinyl halide, 15 to 32 mole % of at least one copolymerizable ethylenically unsaturated hydrocarbon and 3 to 10 mole % of at least one copolymerizable

ethylenically unsaturated monomer containing a carboxyl or amido group; (ii) a film-forming copolymer resin Component B comprising the reaction product of 76 to 94 mole % of at least one copolymerizable vinyl or vinylidene halide, 4 to 12 mole % of acrylonitrile and 2 to 12 mole % of at least one acrylic monomer; and (iii) a film forming mixture of Component A and Component B.

- A formed article according to claim 1 wherein the surfaces are rubber surfaces and the copolymer resin Component A comprises a film-forming copolymer of vinyl halide, alpha-olefin of up to 4 carbon atoms and acrylamide and acrylic acid.
- A formed article according to claim 2 wherein the vinyl halide is vinyl chloride.
- 4. A formed article according to claim 2 wherein the alpha-olefin is ethylene.
- 5. A formed article according to any one of Claims 1-4, wherein each of said inner and outer vinyl surfaces has said slip coating and wherein the slip coating of said outer rubber or vinyl surface comprises component A.
- A formed article according to Claim 5 wherein the slip coating of said outer surface includes an effective amount of a detackifying material.
- 7. A formed article according to Claim 6 wherein the detackifying material is copolymer resin component B of Claim 1.
- 8. A formed article according to Claim 7 wherein the detackifying material is a copolymer of 80 to 90 mole % of vinylidene chloride, 8 to 11 mole % of acrylonitrile, and 2 to 6 mole % of methylmethacrylate.
- A formed article according to Claim 7 wherein the detackifying copolymer is added in an amount of 10 to 25 % by weight (solids basis).
- 10. A formed article according to Claim 6 wherein the detackifying material is a poly(alkylene oxide) polyol.
 - 11. A formed article according to Claim 10 wherein the poly(alkylene oxide) glycol is polypropylene glycol.
 - 12. A formed article according to Claim 5 wherein the first slip coating contains polysiloxane in an amount sufficient to increase the suppleness of the slip coating.
 - 13. A formed article according to Claim 12 wherein the amount of polysiloxane is 0.001 to 0.5 % by weight (solids basis).

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- 14. A formed article according to Claim 5 wherein copolymer resin Component B comprises the film-forming reaction product of 80 to 90 mole % of vinyl halide or vinylidene halide, 8 to 11 mole % of acrylonitrile and 2 to 6 mole % of acrylic monomer.
- 15. A formed article according to Claim 14 wherein the copolymer resin Component B comprises the reaction product of vinylidene chloride, acrylonitrile and methylmethacrylate.
- 16. A formed article according to Claim 5 wherein the second slip coating includes alkali metal, alkaline earth metal, ammonium or trialkanolamine salt of a fatty acid having 8 to 24 carbon atoms in an amount of 2 to 25% by weight (solids basis).
- A formed article according to Claim 16 wherein the salt of the fatty acid is an alkali metal fatty acid salt.
- 18. A formed article according to Claim 17 wherein the alkali metal fatty acid salt is cesium stearate.
- 19. A formed article according to claim 16 wherein the second slip coating includes a poly(alkylene oxide) polyol in an amount of 5 to 50% by weight (solids basis).
- A formed article according to claim 19 wherein the amount of poly(alkylene oxide) polyol is 15 to 30% by weight (solids basis).
- A formed article according to claim 20 wherein the poly(alkylene oxide) polyol is polyethylene glycol.
- A formed article according to claim 20 wherein the poly(alkylene oxide) is polypropylene glycol.
- 23. A formed article according to claim 22 wherein the polypropylene glycol is non-water-soluble polypropylene glycol having a molecular weight about 500 and said polypropylene glycol is emulsified using cesium stearate, potassium stearate or other alkali metal fatty acid.
- 24. A composition comprising aqueous dispersion of copolymer resin Component A of claim 1 and detackifying amounts of Component B of claim 1.
- 25. A composition according to claim 24 further containing a polysiloxane emulsion in an amount sufficient to increase suppleness upon coating a rubber or vinyl surface with said dispersion.
- 26. A formed article according to claim 5 wherein the rubber or vinyl substrate includes a polysiloxane emulsion in an amount sufficient to improve suppleness of the article formed.

- 27. A formed article according to claim 5 wherein Component A or B as the inner or second slip coating includes 2 to 25% by weight (solids basis) of an alkali metal, alkaline earth metal, ammonium or trial-kanolamine salt of a fatty acid of 8 to 24 carbon atoms.
- 28. A formed article according to claim 5 wherein Component A or B further includes 5 to 50% by weight (solids basis) of poly(alkylene oxide) polyol or polyalkyleneglycol ester.
- 29. A formed article according to claim 28 wherein Component A or B includes up to 10% of an emulsified polydimethylsiloxane or aminopolysiloxane.
- 30. An improved process for the production of rubber or vinyl articles having a slip coating on inner and outer surfaces thereof without the need of release agents which comprises coating a form with the aqueous dispersion of copolymer resin Component A of claim 1, forming a rubber or vinyl substrate onto said form, at least partially curing said substrate, coating the surface of the resulting cured substrate with the aqueous dispersion of copolymer resin Component A or B or mixtures thereof of claim 1 and stripping the resulting article from the form.
- A process according to claim 30 wherein the article is a rubber glove.
- A process according to claim 30 wherein the article is a vinyl glove.
- 33. A process according to claim 30 wherein the aqueous dispersion used to coat the form is the aqueous dispersion of claim 25.
- 34. A process according to claim 30 wherein the aqueous dispersion used to coat said rubber surface further comprises 5 to 50% by weight (solids basis) of poly (alkylene oxide) polyol or polyalkyleneglycol ester.
- 35. A process according to claim 34 wherein the aqueous dispersion used to coat said rubber surface further comprises up to 10% of an emulsified polydimethylsiloxane or aminopolysiloxane.

50 Patentansprüche

 Geformter Gegenstand, umfassend äußere und innere Gummi- oder Vinylharzoberflächen, wobei mindestens eine der äußeren und inneren Oberflächen eine im wesentlichen Gummi-freie, dehnbare, anhaftende Gleitbeschichtung aufweist, ausgewählt aus der Gruppe bestehend aus (i) einem ersten filmbildenden Copolymer-Harz-Bestandteil A, umfassend das Reaktionsprodukt aus 55 bis 80 Mol-% von

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mindestens einem copolymerisierbaren Vinylhalogenid, 15 bis 32 Mol-% von mindestens einem copolymerisierbaren ethylenisch ungesättigten Kohlenwasserstoff und 3 bis 10 Mol-% von mindestens einem copolymerisierbaren ethylenisch ungesättigten Monomer, welches eine Carboxyl- oder Amidgruppe enthält; (ii) einem filmbildenden Copolymer-Harz-Bestandteil B, umfassend das Reaktionsprodukt aus 76 bis 94 Mol-% von mindestens einem copolymerisierbaren Vinyl- oder Vinylidenhalogenid, 4 bis 12 Mol-% Acrylnitril und 2 bis 12 Mol-% von mindestens einem Acrylmonomer; und (iii) einem-filmbildenden-Gemisch-des-Bestandteils-A und des Bestandteils B.

- Geformter Gegenstand nach Anspruch 1, worin die Oberflächen Gummioberflächen sind und der Copolymer-Harz-Bestandteil A ein filmbildendes Copolymer aus Vinylhalogenid, alpha-Olefin von bis zu 4 Kohlenstoffatomen und Acrylamid und Acrylsäure

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- Geformter Gegenstand nach Anspruch 2, worin das Vinylhalogenid Vinylchlorid ist.
- 4. Geformter Gegenstand nach Anspruch 2, worin das alpha-Olefin Ethylen ist.
- Geformter Gegenstand nach einem der Ansprüche 1 bis 4, worin jede der inneren und äußeren Vinyloberflächen die Gleitbeschichtung aufweist und worin die Gleitbeschichtung der äußeren Gummioder Vinyloberfläche den Bestandteil A umfaßt.
- Geformter Gegenstand nach Anspruch 5, worin die Gleitbeschichtung der äußeren Oberfläche eine wirksame Menge eines trennenden Materials beinhaftet.
- Geformter Gegenstand nach Anspruch 6, worin das trennende Material der Copolymer-Harz-Bestandteil B von Anspruch 1 ist.
- Geformter Gegenstand nach Anspruch 7, worin das trennende Material ein Copolymer aus 80 bis 90 Mol-% Vinylidenchlorid, 8 bis 11 Mol-% Acrylnitril und 2 bis 6 Mol-% Methylmethacrylat ist.
- Geformter Gegenstand nach Anspruch 7, worin das trennende Copolymer in einer Menge von 10 bis 25 Gew.-% (auf Feststoffbasis) zugegeben ist.
- Geformter Gegenstand nach Anspruch 6, worin das trennende Material ein Poly(alkylenoxid)polyol ist.
- Geformter Gegenstand nach Anspruch 10, worin das Poly(alkylenoxid)glykol Polypropylenglykol ist.

- Geformter Gegenstand nach Anspruch 5, worin die erste Gleitbeschichtung Polysiloxan in einer Menge enthält, die ausreicht, um die Geschmeidigkeit der Gleitbeschichtung zu erhöhen.
- Geformter Gegenstand nach Anspruch 12, worin die Menge von Polysiloxan 0,001 bis 0,5 Gew.-% (auf Feststoffbasis) beträgt.
- 14. Geformter Gegenstand nach Anspruch 5, worin der Copolymer-Harz-Bestandteil B das filmbildende Reaktionsprodukt aus 80 bis 90 Mol-% Vinylhalogenid-oder Vinylidenhalogenid, 8 bis 11 Mol-% Acrylnitril und 2 bis 6 Mol-% Acrylmonomer umfaßt.
 - Geformter Gegenstand nach Anspruch 14, worin der Copolymer-Harz-Bestandteil B das Reaktionsprodukt aus Vinylidenchlorid, Acrylnitril und Methylmethacrylat umfaßt.
 - 16. Geformter Gegenstand nach Anspruch 5, worin die zweite Gleitbeschichtung ein Alkalimetall-, ein Erdalkalimetall-, ein Ammonium- oder Trialkanolaminsalz einer Fettsäure mit 8 bis 24 Kohlenstoffatomen in einer Menge von 2 bis 25 Gew.-% (auf Feststoffbasis) beinhaltet.
 - Geformter Gegenstand nach Anspruch 16, worin das Salz der Fettsäure ein Alkalimetallfettsäuresalz ist
 - Geformter Gegenstand nach Anspruch 17, worin das Alkalimetallfettsäuresalz Cäsiumstearat ist.
 - Geformter Gegenstand nach Anspruch 16, worin die zweite Gleitbeschichtung ein Poly(alkylenoxid)polyol in einer Menge von 5 bis 50 Gew.-% (auf Feststoffbasis) beinhaltet.
- Qu. Geformter Gegenstand nach Anspruch 19, worin die Menge von Poly(alkylenoxid)polyol 15 bis 30 Gew.-% (auf Feststoffbasis) beträgt.
 - Geformter Gegenstand nach Anspruch 20, worin das Poly(alkylenoxid)polyol Polyethylenglykol ist.
 - Geformter Gegenstand nach Anspruch 20, worin das Poly(alkylenoxid) Polypropylenglykol ist.
- 23. Geformter Gegenstand nach Anspruch 22, worin das Polypropylenglykol ein nicht wasserlösliches Polypropylenglykol ist mit einem Molekulargewicht von ungefähr 500 und worin das Polypropylenglykol unter Verwendung von Cäsiumstearat, Kaliumstearat oder einer anderen Alkalimetallfettsäure emulgiert ist.
 - 24. Zusammensetzung, umfassend eine wäßrige Dispersion des Copolymer-Harz-Bestandteils A von

- 25. Zusammensetzung nach Anspruch 24, weiterhin enthaltend eine Polysiloxan-Emulsion in einer 5 Menge, die ausreicht, um die Geschmeidigkeit bei Beschichten einer Gummi- oder Vinyloberfläche mit der Dispersion zu erhöhen.
- 26. Geformter Gegenstand nach Anspruch 5, worin das Gummi- oder Vinylsubstrat eine Polysiloxan-Emulsion beinhaltet in einer Menge, die ausreicht, um die Geschmeidigkeit des geformten Gegenstandes zu verbessern.
- 27. Geformter Gegenstand nach Anspruch 5, worin der Bestandteil A oder B als die innere oder zweite Gleitbeschichtung 2 bis 25 Gew.-% (auf Feststoffbasis) eines Alkalimetall-, Erdalkalimetall-, Ammoniumoder Trialkanolaminsalzes einer Fettsäure mit 8 bis 24 Kohlenstoffatomen beinhaltet.
- 28. Geformter Gegenstand nach Anspruch 5, worin der Bestandteil A oder B weiterhin 5 bis 50 Gew.-% (auf Feststoffbasis) Poly(alkylenoxid)polyol oder Polyalkylenglykolester beinhaltet.
- 29. Geformter Gegenstand nach Anspruch 28, worin der Bestandteil A oder B bis zu 10 % eines emulgierten Polydimethylsiloxans oder Aminopolysiloxans beinhaltet.
- 30. Verbessertes Verfahren zur Herstellung von Gummi- oder Vinylgegenständen mit einer Gleitbeschichtung auf inneren und äußeren Oberflächen davon ohne die Notwendigkeit von Trennmitteln, welcher umfaßt Beschichten einer Form mit der wäßrigen Dispersion des Copolymer-Harz-Bestandteils A von Anspruch 1, Bilden eines Gummi- oder Vinylsubstrats auf der Form, mindestens teilweises Aushärten des Substrats, Beschichten der Oberfläche des resultierenden gehärteten Substrats mit der wäßrigen Dispersion des Copolymer-Harz-Bestandteils A oder B oder Gemischen davon von Anspruch 1 und Abziehen des resultierenden Gegenstandes von der Form.
- Verfahren nach Anspruch 30, worin der Gegenstand ein Gummihandschuh ist.
- Verfahren nach Anspruch 30, worin der Gegenstand ein Vinylhandschuh ist.
- 33. Verfahren, nach Anspruch 30, worin die zum Beschichten der Form verwendete w\u00e4\u00dfrige Dispersion die w\u00e4\u00dfrige Dispersion nach Anspruch 25 ist.
- Verfahren nach Anspruch 30, worin die zum Beschichten der Gummioberfläche verwendete

wäßrige Dispersion weiterhin 5 bis 50 Gew.-% (auf Feststoffbasis) Poly(alkylenoxid)Polyol oder Polyethylenglykolester umfaßt.

35. Verfahren nach Anspruch 34, worin die zum Beschichten der Gummioberfläche verwendete wäßrige Dispersion weiterhin bis zu 10 % eines emulgierten Polydimethylsiloxans oder Aminopolysiloxans umfaßt.

Revendications

- 1. Article faconné constitué d'une surface extérieure et d'une surface intérieure de caoutchouc ou de résine vinylique, au moins une desdites surfaces extérieure et intérieure ayant un revêtement autoglissant, adhérent, extensible, essentiellement dépourvu de caoutchouc, choisi dans le groupe constitué de (i) un premier composant A fait d'une résine copolymère filmogène comprenant le produit de la réaction de 55 à 80 % molaires d'au moins un halogénure de vinyle copolymérisable, 15 à 32 % molaires d'au moins un hydrocarbure à insaturation éthylénique copolymérisable et 3 à 10 % molaires d'au moins un monomère à insaturation éthylénique copolymérisable contenant un groupe carboxy ou amido ; (ii) un composant B fait d'une résine copolymère filmogène comprenant le produit de la réaction de 76 à 94 % molaires d'au moins un halogénure de vinyle ou de vinylidène copolymérisable, 4 à 12 % molaires d'acrylonitrile et 2 à 12 % molaires d'au moins un monomère acrylique ; et (iii) un mélange filmogène du composant A et du composant B.
- 35 2. Article façonné selon la revendication 1, dans lequel les surfaces sont des surfaces de caoutchouc et le composant A fait d'une résine copolymère comprend un copolymère filmogène d'halogénure de vinyle, d'une α-oléfine ayant jusqu'à 4 atomes de carbone et d'acrylamide et d'acide acrylique.
 - Article façonné selon la revendication 2, dans lequel l'halogénure de vinyle est le chlorure de vinyle.
- 45 4. Article façonné selon la revendication 2, dans lequel l'α-oléfine est l'éthylène.
 - 5. Article façonné selon l'une quelconque des revendications 1 à 4, dans lequel chacune desdites surfaces vinyliques intérieure et extérieure a ledit revêtement autoglissant et dans lequel le revêtement autoglissant de ladite surface extérieure de caoutchouc ou de vinyle comprend le composant A.
 - Article façonné selon la revendication 5, dans lequel le revêtement autoglissant de ladite surface extérieure comprend une quantité efficace d'une matière réduisant le collant.

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- 7. Article façonné selon la revendication 6, dans lequel la matière réduisant le collant est le composant B fait d'une résine copolymère de la revendication 1.
- Article façonné selon la revendication 7, dans lequel 5 la matière réduisant le collant est un copolymère de 80 à 90 % molaires de chlorure de vinylidène, 8 à 11 % molaires d'acrylonitrile et 2 à 6 % molaires de méthacrylate de méthyle.
- Article façonné selon la revendication 7, dans lequel le copolymère réduisant le collant est ajouté en une proportion de 10 à 25 % en poids (en matières sèches).
- Article façonné selon la revendication 6, dans lequel la matière réduisant le collant est un poly(oxyde d'alkylène)polyol.
- Article façonné selon la revendication 10, dans lequel le poly(oxyde d'alkylène)glycol est le polypropylèneglycol.
- 12. Article façonné selon la revendication 5, dans lequel le premier revêtement autoglissant contient un polysiloxane en une quantité suffisante pour accroître la souplesse du revêtement autoglissant.
- Article façonné selon la revendication 12, dans lequel la quantité du polysiloxane est de 0,001 à 0,5
 % en poids (en matières sèches).
- 14. Article façonné selon la revendication 5, dans lequel le composant B fait d'une résine copolymère comprend le produit réactionnel filmogène de 80 à 90 % molaires d'halogénure de vinyle ou d'halogénure de vinylidène, 8 à 11 % molaires d'acrylonitrile et 2 à 6 % molaires de monomère acrylique.
- 15. Article façonné selon la revendication 14, dans lequel le composant B fait d'une résine copolymère comprend le produit réactionnel du chlorure de vinylidène, de l'acrylonitrile et du méthacrylate de méthyle.
- 16. Article façonné selon la revendication 5, dans lequel le second revêtement autoglissant comprend un sel de métal alcalin, de métal alcalino-terreux, d'ammonium ou de trialcanolamine d'un acide gras ayant 8 à 24 atomes de carbone en une proportion de 2 à 25 % en poids (en matières sèches).
- Article façonné selon la revendication 16, dans lequel le sel de l'acide gras est un sel de métal alcalin d'acide gras.
- Article façonné selon la revendication 17, dans lequel le sel de métal alcalin d'acide gras est le stéarate de césium.

- 19. Article façonné selon la revendication 16, dans lequel le second revêtement autoglissant comprend un poly(oxyde d'alkylène)polyol en une proportion de 5 à 50 % en poids (en matières sèches).
- Article façonné selon la revendication 19, dans lequel la proportion du poly(oxyde d'alkylène)polyol est de 15 à 30 % en poids (en matières sèches).
- 21. Article façonné selon la revendication 20, dans lequel le poly(oxyde d'alkylène)polyol est le polyéthylèneglycol.
 - Article façonné selon la revendication 20, dans lequel le poly(oxyde d'alkylène)polyol est le polypropylèneglycol.
 - 23. Article façonné selon la revendication 22, dans lequel le polypropylèneglycol est un polypropylèneglycol insoluble dans l'eau ayant un poids moléculaire d'environ 500 et ledit polypropylèneglycol est émulsifié par utilisation de stéarate de césium, de stéarate de potassium ou d'un autre sel de métal alcalin d'acide gras.
 - 24. Composition comprenant une dispersion aqueuse du composant A fait d'une résine copolymère de la revendication 1 et de quantités réduisant le collant du composant B de la revendication 1.
 - 25. Composition selon la revendication 24, comprenant de plus une émulsion de polysiloxane en une quantité suffisante pour accroître la souplesse lorsqu'on revêt une surface de caoutchouc ou de vinyle avec ladite dispersion.
 - 26. Article façonné selon la revendication 5, dans lequel le substrat de caoutchouc ou de vinyle comprend une émulsion de polysiloxane en une quantité suffisante pour améliorer la souplesse de l'article façonné.
 - 27. Article façonné selon la revendication 5, dans lequel le composant A ou B, en tant que revêtement autoglissant intérieur ou en tant que second revêtement autoglissant, comprend 2 à 25 % en poids (en matières sèches) d'un sel de métal alcalin, de métal alcalino-terreux, d'ammonium ou de trialcanolamine d'un acide gras de 8 à 24 atomes de carbone.
 - 28. Article façonné selon la revendication 5, dans lequel le composant A ou le composant B comprend de plus 5 à 50 % en poids (en matières sèches) de poly(oxyde d'alkylène)polyol ou d'un ester de polyalkylèneglycol.
 - 29. Article façonné selon la revendication 28, dans lequel le composant A ou le composant B comprend

jusqu'à 10 % d'un polydiméthylsiloxane ou d'un aminopolysiloxane émulsifiés.

- 30. Procédé perfectionné pour la production d'articles de caoutchouc ou de vinyle ayant un revêtement autoglissant sur leurs surfaces intérieure et extérieure, sans nécessiter d'agents de démoulage, qui comprend le revêtement d'une forme avec une dispersion aqueuse du composant A fait d'une résine de copolymère de la revendication 1, la formation d'un substrat de caoutchouc ou de vinyle sur ladite forme, le durcissement au moins partiel dudit substrat, le revêtement de la surface du substrat durci obtenu avec une dispersion aqueuse du composant A ou du composant B fait d'une résine copolymère ou de leurs mélanges de la revendication 1 et la séparation de l'article obtenu d'avec la forme.
- Procédé selon la revendication 30, dans lequel l'article est un gant de caoutchouc.
- 32. Procédé selon la revendication 30, dans lequel l'article est un gant de vinyle.
- 33. Procédé selon la revendication 30, dans lequel la 25 dispersion aqueuse utilisée pour revêtir la forme est la dispersion aqueuse de la revendication 25.
- 34. Procédé selon la revendication 30, dans lequel la dispersion aqueuse utilisée pour revêtir ladite surface de caoutchouc comprend de plus 5 à 50 % en poids (en matières sèches) de poly(oxyde d'alkylène)polyol ou d'ester de polyéthylèneglycol.
- 35. Procédé selon la revendication 34, dans lequel la dispersion aqueuse utilisée pour revêtir ladite surface de caoutchouc comprend de plus jusqu'à 10 % d'un polydiméthylsiloxane ou d'un aminopolysiloxane émulsifiés.

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